

NOVEL CYTOTOXIC RING-A *SECO*-CYCLOARTANE TRITERPENES FROM *GARDENIA CORONARIA* AND *G. SOOTEPENSIS*

Gloria L. Silva, Roberto R. Gil, Baoliang Cui, Heebyung Chai, Thawatchai Santisuk,¹ Ekarath Srisook,²
Vichai Reutrakul,² Patoomratana Tuchinda,² Smaisukh Sophasan,³ Suparp Sujarit,³ Suchart Upatham,⁴
Sean M. Lynn,⁵ John E. Farthing,⁵ Shi-Lin Yang,⁵ Jane A. Lewis,⁵ Melanie J. O'Neill,⁵
Norman R. Farnsworth, Geoffrey A. Cordell, John M. Pezzuto, and A. Douglas Kinghorn*

Program for Collaborative Research in the Pharmaceutical Sciences, Department of Medicinal Chemistry
and Pharmacognosy, College of Pharmacy, University of Illinois at Chicago, Chicago, IL 60612, U.S.A.;

¹Royal Forest Herbarium, Bangkok 10900, Thailand; Department of ²Chemistry, ³Physiology, and

⁴Biology, Mahidol University, Bangkok 10400, Thailand; ⁵Glaxo Wellcome Medicines Research Centre,
Gunnels Wood Road, Stevenage, Herts., SG1 2NY, U.K.

Abstract: Coronalolide methyl ester (**1**), coronalolide (**2**), and coronalolic acid (**3**) were isolated from the leaves and/or stems of *Gardenia coronaria*. A further compound, methyl coronalolate acetate (**4**), was purified from the stems after methylation. The novel compounds **1-4** have the rare ring-A *seco*-cycloartane carbon skeleton and their structures were assigned on the basis of spectral data and molecular modeling, as well as X-ray crystallography performed on **1**. Compounds **1** and **2** were also isolated from the leaves of *Gardenia sootepensis* and showed broad cytotoxic activity when evaluated against a panel of human cancer cell lines. Copyright © 1996 Elsevier Science Ltd

INTRODUCTION

The genus *Gardenia* (Rubiaceae) comprises more than 80 species with several being used ethnomedically in various countries primarily for abortifacient (*G. griffithii*¹⁾) and contraceptive (*G. jasminoides*,¹⁾ *G. jovis-tonantis*,²⁾ *G. turgida*³⁾) purposes. Other uses are as a febrifuge (*G. laeta*),⁴⁾ for the treatment of headaches (*G. taitensis*⁵⁾), and as a larvicide (*G. campanulata*).⁶⁾ Extracts of various *Gardenia* species have shown anti-implantation and abortifacient effects,⁷⁾ as well as antibacterial,⁸⁾ antiulcer,⁹⁾ analgesic,¹⁰⁾ diuretic,¹⁰⁾ hypotensive,¹⁰⁾ and larvicidal activity.¹¹⁾ In our ongoing project on the discovery of new anticancer agents from plants,¹²⁾ we have studied the chloroform extracts of both the stems and leaves of *G. coronaria* (Buch.-Ham.) and the leaves of *G. sootepensis* Hutch., both species for which there appears to be no previous chemical literature. In the present work, we report the isolation and structure elucidation of four novel *seco*-cycloartanes (**1-4**), of which compounds **1** and **2** showed significant cytotoxic effects against a panel of human cancer cell lines, along with two known compounds, 5,7,3'-trihydroxy-6,4',5'-trimethoxyflavone and benzoic acid. This is the first report of ring-A 3,4-*seco*-cycloartanes in the genus

Gardenia; to our knowledge only a small group of such compounds have been isolated from natural sources, including nigranoic acid,^{13,14} schisanlactone B,¹⁵ 24-methylene-3,4-*seco*-cycloart-4(28)-en-3-oic acid,¹⁶ kadsulactone A,¹⁷ kadsudilactone,¹⁸ pseudolarolide C,¹⁹ and several constituents from *Tillandsia usneoides*.²⁰ Compounds 1-4 and compounds 1-2 were isolated from *G. coronaria* and *G. sootepensis*, respectively.

RESULTS AND DISCUSSION

Compound 1 was shown to possess a molecular formula of C₃₁H₄₄O₅ by HREIMS. Its IR spectrum showed three carbonyl group absorbances at 1768, 1739 and 1684 cm⁻¹, indicating the presence of two ester or lactone groups and one conjugated carbonyl group. The ¹H-NMR spectrum of 1 exhibited a characteristic pair of doublets at δ 0.43 (5.5 Hz) and δ 0.19 (5.5 Hz), corresponding to the C-19 methylene protons of the cyclopropane ring of a cycloartane triterpene.¹³⁻²⁰ The signal at δ 9.40 (1H, s) confirmed the presence of an aldehyde group, which, together with a triplet of quartets at δ 6.49 (*J*=6.5, 1.5 Hz, H-24), a three-proton doublet at 0.94 (H₃-21), and a broad three-proton singlet at δ 1.76 (H₃-27), indicated the presence of a side-chain attached to C-17. This was confirmed by the EIMS fragment ion at *m/z* 371, consistent with a cleavage between C-17 and C-20.

The above data suggested initially that compound 1 might be a normal cycloartane-type triterpene; however, the absence of signals corresponding to *gem*-dimethyl groups at C-4, the presence of two more carbonyl peaks (either ester or lactone) in the IR spectrum, and the analysis of a combination of the ¹H- (Table 1), ¹³C- (Table 2) (broad-band and DEPT), DQF-COSY, HETCOR, HOHAHA, ROESY, and selective INEPT (Table 3) NMR spectra, led to the proposal of structure 1 for this compound. This structure is substantially different from those ring-A *seco*-cycloartanes so far isolated,¹³⁻²⁰ since C-4 was lactonized to C-6, showing a similar structural characteristic to the sesquiterpene lactones.²¹ Two doublets at δ 6.34 (2.5 Hz) and δ 5.75 (2.3 Hz) in the ¹H-NMR spectrum of 1 were assigned to H-28a and H-28b of the exomethylene γ-lactone ring. Using these signals as a starting point in the DQF-COSY spectrum, signals were assigned to H-5 (δ 3.23), H-6 (δ 4.74), H-7α (δ 1.77), H-7β (δ 1.55), and H-8 (δ 2.12). Three methyl groups appeared as singlets (H₃-18, H₃-27, H₃-30), and another as a doublet at δ 0.94 (6.5 Hz, H₃-21). From the latter signal, the DQF-COSY spectrum was used to assign the resonance corresponding to H-20 at δ 1.48, which did not overlap any other signal. Following the cross-peaks from H-20, the resonances of H-22a and H-22b could be found at δ 1.63 and δ 1.24, respectively. An additional cross-peak was found at *ca.* δ 1.65, and assigned to H-17. This signal showed cross-peaks with the resonances at δ 1.95 (m) and δ 1.37 (m), attributable to H-16a and H-16b, respectively. No further correlations were observed between these signals and those corresponding to H-15a,b. Both H-15 proton resonances were found at the same chemical shift, δ 1.37. Most previous structure elucidation arguments for cycloartane derivatives have been based on data derived from ¹³C-NMR

spectroscopy and there are only a few reports on the use of high-field $^1\text{H-NMR}$ studies (500-600 MHz).²²⁾ There was another group of correlations from the signals corresponding to H-22a and H-22b with the resonances of H-23a and H-23b (δ 2.43 and δ 2.32), the resonance of H-24 at δ 6.49, and the broad singlet of H₃-27 at δ 1.76 (Table 1). The signal of H-24 showed a nOe cross-correlation peak in the ROESY experiment with the resonance of the aldehyde group at δ 9.40 (H-26), indicating an *E*-configuration of the double bond between C-24 and C-25. The HOHAHA experiment exhibited a correlation peak between H-20 and both the H-16a and H-16b signals, due to a relay through H-17. A double-relay cross-peak was observed between H-22a and H-16b.

Table 1. $^1\text{H-NMR}$ data of compounds 1-4 (CDCl_3).*

H#	1 ^a	2 ^b	3 ^b	4 ^a
1a	2.25 obsc.	--	--	--
1b	1.61 obsc.	--	--	--
2a	2.55 obsc.	--	--	--
2b	2.45 obsc.	--	--	--
5	3.23 (bd, 8.0)	3.25 (bd, 8.4)	--	--
6	4.74 (td, 8.0, 6.5)	4.77 (td, 8.4, 6.7)	--	--
7 α	1.77 obsc.	--	--	--
7 β	1.55 obsc.	--	--	--
8	2.12 (t, 5.5)	--	--	--
11 } 12 }	1.6-1.7	--	--	--
15a	1.37 obsc.	--	--	--
15b	1.37 obsc.	--	--	--
16a	1.95 obsc.	--	--	--
16b	1.37 obsc.	--	--	--
17	~1.65 obsc.	--	--	--
18	0.95 (s)	0.95 (s)	0.98 (s)	0.97 (s)
19 β	0.43 (d, 5.5)	0.44 (d, 5.4)	0.74 (d, 4.2)	0.72 (d, 5)
19 α	0.19 (d, 5.5)	0.19 (d, 5.4)	0.50 (d, 4.2)	0.49 (d, 5)
20	1.48 (m)	--	--	--
21	0.94 (d, 6.5)	0.94 (d, 6.5)	0.93 (d, 6.5)	0.93 (d, 6.5)
22a	1.63 obsc.	--	--	--
22b	1.24 obsc.	--	--	--
23a	2.43 obsc.	--	--	--
23b	2.32 obsc.	--	--	--
24	6.49 (tq, 6.5, 1.5)	6.51 (bt, 6.2)	6.51 (bt, 6.8)	6.49 (bt, 6.5)
26	9.40 (s)	9.40 (s)	9.40 (s)	9.40 (s)
27	1.76 (bs)	1.76 (bs)	1.78 (bs)	1.76 (bs)
28a	6.34 (d, 2.5)	6.35 (d, 2.2)	5.11 (s)	5.14 (s)
28b	5.75 (d, 2.3)	5.75 (d, 2.0)	5.09 (s)	5.09 (s)
29	--	--	4.12 (bs)	4.57 (bs)
30	0.92 (s) 3.69 (s, OMe)	0.92 (s) --	0.94 (s) --	0.94 (s) 2.10 (s, Ac) 3.65 (s, OMe)

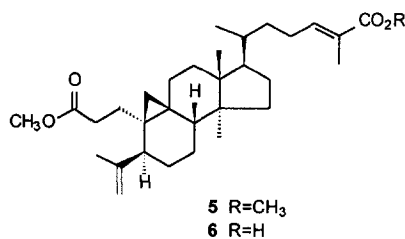
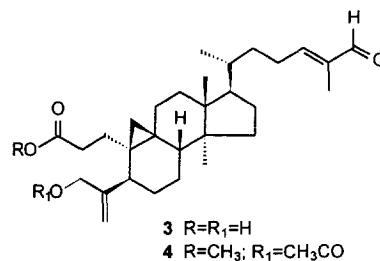
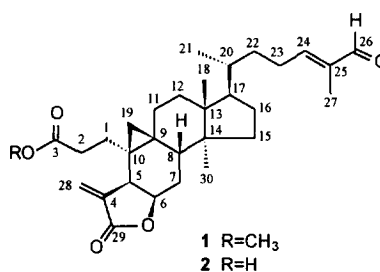


Figure 1. Structures of compounds 1-6.

* Chemical shifts given in ppm using TMS as internal reference.

Multiplicities and coupling constants (Hz) are given in parentheses.

^a Obtained at 500 MHz. ^b Obtained at 300 MHz. Obsc.=Obscured signals.

We assumed initially that the stereochemistry at C-5, C-8, C-9, C-10, C-13, C-14, C-17, and C-20 in **1** was the same as that long-established for the cycloartanes. The stereochemistry was determined using molecular modeling and a ROESY experiment. The structures of both epimers at C-6 (*cis* and *trans* fusion of the γ -lactone ring) were generated by the molecular modeling software PCMODEL V 5.1 for Windows using the MMX force field. From the calculated dihedral angles involving H-5, H-6 and H₂-7, we have estimated vicinal coupling constant using the Karplus generalized equation developed by Haasnoot *et al.*²³⁾ The observed vicinal coupling constants involving H-5, H-6 and H₂-7 were very similar to those calculated for the model of the *cis*-isomer (Table 4). In the ROESY experiment H-19 α (δ 0.43) showed NOE cross-correlation peaks with the signals corresponding to H-7 β (δ 1.55) and H-8 (δ 2.12); and the signals of H-5 and H-6 showed nOe

Table 2. ¹³C-NMR data of compounds **1-3**, and **5**.*

C#	1	2	3	5 [†]
1	30.8	30.7	28.1	29.1
2	31.1	31.2	30.3	31.5
3	173.3	178.1	179.3	174.4
4	139.0	139.1	152.1	149.5
5	38.9	39.0	41.9	45.9
6	74.3	74.4	28.7	27.8
7	27.1	27.2	25.2	25.1
8	38.2	38.2	47.9	47.8
9	25.0	25.1	21.8	21.4
10	28.1	28.1	27.4	27.2
11	26.4	26.5	26.9	27.1
12	32.9	33.0	33.0	33.1
13	45.7	45.7	45.1	45.3
14	48.6	48.6	48.9	49.0
15	34.7	34.8	35.6	35.7
16	27.7	27.8	28.1	28.1
17	51.2	51.4	52.1	52.2
18	15.9	16.0	18.1 ^a	18.1 ^a
19	23.1	23.2	30.3	30.0
20	35.9	36.0	36.0	36.0
21	18.2	18.2	18.2 ^a	18.2 ^a
22	34.7	34.8	34.7	35.2
23	25.9	26.0	26.1	25.7
24	155.2	155.4	155.7	143.2
25	139.0	139.1	139.2	127.2
26	195.2	195.5	196.0	168.8
27	9.1	9.2	9.2	12.4
28	123.0	123.1	110.6	111.6
29	170.6	170.7	64.7	19.4
30	20.0	20.1	19.4	19.8
OMe	51.7			51.5, 51.7

* Chemical shifts given in ppm using TMS as internal reference in CDCl₃ (90.8 MHz).

^a Interchangeable assignments.

[†] ¹³C-NMR data from literature.¹⁸⁾

Table 3. Selective INEPT NMR data for **1** (CDCl₃).

Proton (s) irradiated	Carbon (s) enhanced
H-5	23.1 (C-19), 27.1 (C-7), 28.1 (C-10) 30.8 (C-1), 74.3 (C-6), 123.0 (C-28) 139.02 (C-4), 170.6 (C-29)
H-6	28.1 (C-10), 38.2 (C-8), 139.02 (C-4) 170.6 (C-29)
H-8	20.0 (C-30), 23.1 (C-19), 25.0 (C-9) 27.1 (C-7), 28.1 (C-10), 34.66 (C-15) 45.7 (C-13), 48.6 (C-14), 74.3 (C-6)
H ₃ -18, 21, 30	32.9 (C-12), 34.66 (C-15, C-22) 35.9 (C-20), 38.2 (C-8), 45.7 (C-13) 48.6 (C-14), 51.2 (C-17)
H-19 α	25.0 (C-9), 28.1 (C-10), 30.8 (C-1) 38.2 (C-8), 38.9 (C-5)
H-19 β	25.0 (C-9), 28.1 (C-10), 30.8 (C-1) 38.2 (C-8), 38.9 (C-5), 26.4 (C-11)
H-24	9.1 (C-27), 34.66 (C-22), 25.9 (C-23) 195.2 (C-26)
H-26	138.97 (C-25)
H-28a	38.9 (C-5), 139.02 (C-4), 170.6 (C-29)
H-28b	38.9 (C-5), 170.6 (C-29)

Table 4. Observed vs. calculated vicinal *J* values (Hz) involving H-5, H-6 and H₂-7 in compound **1**.

<i>J</i> (Hz)	Calculated		Observed
	<i>cis</i> -isomer	<i>trans</i> -isomer	1
H-5, H-6	8.99	10.85	8.0
H-6, H-7 α	7.34	10.11	6.5
H-6, H-7 β	8.48	5.67	8.0

cross-correlation peaks with the three-proton singlet at δ 0.92, which was consequently assigned to H₃-30. These results again agreed very well with the spatial relationship observed for these protons in the model of the *cis*-isomer. The signal of H-8 also showed a cross-peak with the resonance of H₃-18 (δ 0.95, s), and the signal of H-7 α showed a cross-peak correlation with the singlet corresponding to H₃-30. The relative stereochemistry of this compound was confirmed by single-crystal X-ray diffraction analysis, and a three-dimensional view of the structure is given in Figure 2.

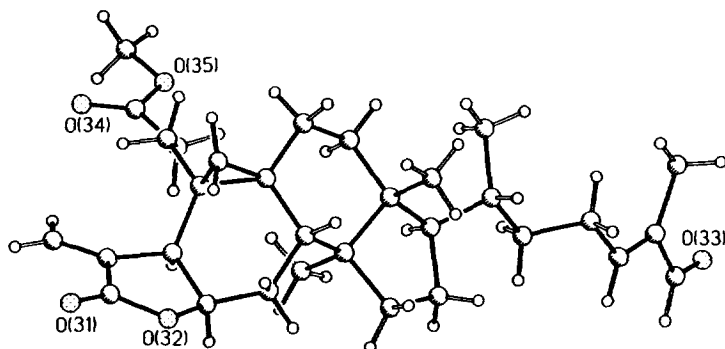


Figure 2. Perspective view of the molecular structure of **1** as determined by X-ray crystallography.

The ¹³C-NMR (Table 2) signals of **1** were fully assigned using the DEPT, HETCOR and selective INEPT experiments, and also by comparison with the published ¹³C-NMR data of cycloartanes. The selective INEPT experiment results (Table 3) were helpful in the assignment of the quaternary carbons and also allowed the differentiation of the resonances of C-4 and C-25, which varied by only 0.05 ppm. Irradiation of the aldehyde signal at δ 9.40 (H-26) produced an enhancement of the signal at 138.97 (C-25), while selective polarization transfer from the resonances of H-5, H-6 and H-28a enhanced the signal at δ 139.02 (C-4). The remaining irradiations presented in Table 3 substantiated the proposed structure for coronalolide methyl ester (**1**). Assignments for the ¹H-NMR signals corresponding to protons H₂-1 and H₂-2 are in agreement with analogous data reported for dimethyl 3,4-*seco*-mangiferonate from *Tillandsia usneoides*.²⁴

Compound **2** revealed a molecular formula of C₃₀H₄₂O₅ by HREIMS, representing one carbon and two hydrogen atoms less than compound **1**. An EIMS fragment peak also showed the loss of 125 Da from the molecular ion at *m/z* 482, indicating the presence of the same side-chain attached to C-17 as in **1**. The IR spectrum of **2** lacked the band at 1739 cm⁻¹ corresponding to the saturated ester at C-3 and exhibited new bands at 1707 cm⁻¹ and 3221 cm⁻¹ (broad shoulder), suggesting the presence of a free carboxylic acid group. The ¹H- and ¹³C-NMR spectra (Tables 1 and 2) of **1** and **2** were almost superimposable, except that the signals corresponding to the ester OMe group were lacking, and the signal due to the carbonyl group at C-3 in the ¹³C-NMR spectrum of coronalolide (**2**) was shifted downfield *ca.* 5 ppm with respect to compound **1**. Hydrolysis

of compound **1** with LiOH in refluxing THF afforded **2** after purification by prep. TLC using 40% CH₂Cl₂-MeOH as solvent system.

Compound **3** demonstrated a molecular formula of C₃₀H₄₆O₄ by HREIMS, representing one oxygen atom and two degrees of unsaturation less than compound **2**. The IR spectrum exhibited aldehyde (1688 cm⁻¹) and free carboxylic acid (1707 cm⁻¹) carbonyl groups, however, as well as hydroxyl group absorption (3431 cm⁻¹). The ¹H- and ¹³C-NMR spectra of **3** (Tables 1 and 2) revealed the presence of the same side-chain as in compounds **1** and **2**. The signal corresponding to the carbonyl of the γ -lactone ring in **1** and **2** was absent, as was the ¹H-NMR signal corresponding to the γ -lactone carbinol proton H-6 of **1** and **2**. Two one-proton, broad singlets at δ 5.09 and δ 5.11 suggested the presence of a terminal double bond. This signal exhibited allylic coupling with the two-proton, broad singlet at δ 4.12. The pair of doublets corresponding to H-19a (δ 0.50) and H-19b (δ 0.74) appeared shifted downfield with respect to analogous signals in compounds **1** and **2**. These data suggested a change in the substitution pattern of ring B, leading to the proposal for coronalolic acid (**3**) of a 3,4-*seco*-cycloartane structure with C-29 oxidized to a primary alcohol.

The ¹³C-NMR DEPT spectrum of **3** (Table 2) revealed the presence of a carbonyl doublet (aldehyde), a carbonyl singlet, four *sp*² carbons (two C, one CH and one CH₂), and twenty-four *sp*³ carbons (four C, four CH, twelve CH₂ and four CH₃). One of these methylenes resonated at δ 64.7, confirming the presence of a primary alcohol group. The ¹³C-NMR data correlated closely with those reported for the 3,4-*seco*-cycloartanes **5**²⁴) and **6**²⁰) isolated from *Tillandsia usneoides*, which differ from compound **3** in the side-chain (methyl ester or free acid instead of aldehyde), at C-3 (methyl ester instead of free acid), and at C-29 (alcohol instead of methyl group). It can be seen in Table 2 that oxidation of the methyl group C-29 to an alcohol produced a γ -effect at C-5 (δ 45.9 in **5** and **6**, δ 41.9 in **3**). Also, a δ -effect was observed in the resonance of C-6, which was shifted slightly downfield in coronalolic acid (**3**) with respect to **5** and **6** (Table 2).

Finally, a very small amount of a fourth 3,4-*seco*-cycloartane, **4**, was purified from a fraction treated with diazomethane. Its CIMS exhibited a quasimolecular ion peak ([M+1]⁺) at *m/z* 527, corresponding to a molecular formula of C₃₃H₅₀O₅. The base peak was observed at *m/z* 467, produced by the loss of acetic acid from the quasimolecular ion. Due to the paucity of **4**, we were only able to obtain a ¹H-NMR spectrum (Table 1). Analysis of the data, however, enabled compound **4** to be proposed as the C-29-acetyl, C-3-methyl ester derivative of **3**, presumably derived from the corresponding acid, **3**.

Compounds **1-3** were evaluated for their cytotoxic effects against a panel of human cancer cell lines.²⁵ While coronalolic acid (**3**) was inactive in all cell lines tested, coronalolide methyl ester (**1**) and coronalolide (**2**) were broadly cytotoxic with ED₅₀ values in the range of 0.1-1.0 μ g/ml [most potent activity for **1**, ED₅₀ 0.6 μ g/ml against a hormone-dependent breast cancer cell line (ZR-75-1); most potent activity for **2**, ED₅₀ 0.5 μ g/ml against a glioma cell line (U373)]. Based on the inactivity of **3**, the cytotoxicity of **1** and **2** appears to be facilitated by the exomethylene lactone ring portion of each molecule.

EXPERIMENTAL

General procedures: Mps: uncorr.; UV: MeOH; IR: film; ^1H - and ^{13}C -NMR spectra were recorded on 300, 360 or 500 MHz instruments with TMS as int. standard; low- and high-resolution mass spectra were obtained on a Finnigan MAT-90 instrument.

Plant material: The leaves and stems of *Gardenia coronaria* Buch.-Ham. (Rubiaceae) and the leaves of *G. sootepensis* Hutch. were collected at Doi Suthep, Chiangmai, Thailand in September, 1993 and in February, 1990, respectively, and identified by one of us (T.S.). The voucher specimens for *G. coronaria* (BKF 93742) and *G. sootepensis* (BKF 101411) have been deposited at the Herbarium, Royal Forestry Department, Bangkok, Thailand.

Extraction and isolation: The dried, powdered leaves of *G. coronaria* (35 g) were extracted by maceration with CHCl_3 three times and the residue was concentrated under vacuum to afford a gum (3.74 g). The dried CHCl_3 extract was absorbed on Si gel (8 g) and fractionated by flash CC (column i.d 2.5 cm packed with Si gel, 15 cm deep) and eluted with hexanes, hexanes-toluene (1:1), toluene, toluene- CHCl_3 (1:1), CHCl_3 , CHCl_3 -EtOAc mixtures of increasing polarity, EtOAc, EtOAc-MeOH (1:1), and MeOH. Fraction 7 (293 mg, CHCl_3 -EtOAc 4%) showed a main UV-active spot, that appeared pink when sprayed with H_2SO_4 /vanillin, and was purified by flash CC with silica H, eluting with CHCl_3 , EtOAc, and MeOH in gradients. Subfractions 3 and 4 were combined (59.8 mg) and coronalolide methyl ester (**1**, 9.0 mg, 0.025%) was purified by prep. TLC with CHCl_3 -EtOAc (9:1) as solvent. From fraction 8 (185 mg), coronalolide (**2**, 12.5 mg, 0.036%) was separated by a combination of normal and C_{18} -reversed phase CC. Fractions 14-16 (207.8 mg, CHCl_3 -EtOAc 70%) were fractionated by flash CC eluted with CHCl_3 and CHCl_3 -MeOH (1, 2, 4, 8 and 12%) and washed with MeOH. Subfraction 13 (62.4 mg) showed a main UV active spot with R_f 0.75 using CHCl_3 -MeOH (85:15), and coronalolic acid (**3**, 9.5 mg, 0.027%) was purified by prep. TLC using this solvent system.

The powdered stems of *G. coronaria* (1.5 kg) were extracted with CHCl_3 (3 x 5 L) at room temperature, filtered, and concentrated under vacuum, giving a dried extract (75.17 g) which, in turn, was fractionated by VLC, eluting with the same solvent system (1.5 L) as for the leaf extract. Fraction 5 (7.01 g, CHCl_3 -EtOAc 5%) was subfractionated by gravity CC (Si gel, 450 g) using as solvent systems hexane, hexane-EtOAc mixtures of increasing polarity, EtOAc-MeOH (19:1 and 9:1), and MeOH. Subfraction 14 (1.31 g) was purified by CC with CHCl_3 to afford 550 mg of **1**. Fraction 6 (15.08 g, CHCl_3 -EtOAc 10%) was fractionated by VLC, and subfraction 6 (2.35 g, hexane-EtOAc 25%) after being purified by CC with CHCl_3 , afforded a further 1.2 g of compound **1**. Fraction 7 (3.67 g, CHCl_3 -EtOAc 15%) was purified by CC (180 g Si gel) using CHCl_3 , and CHCl_3 -MeOH mixture of increasing polarity, and MeOH for elution. Subfractions 5-16 were combined affording a yellow precipitate (126.5 mg, 0.008%) identified as 5,7,3'-trihydroxy-6,4',5'-

trimethoxyflavone.²⁶⁾ Fraction 8 (2.69 g, CHCl₃-EtOAc 20%) was purified by VLC, subfraction 4 affording crystals that purified by sublimation that were identified as benzoic acid²⁷⁾ (2 mg). Fraction 10 (4.13 g) was fractionated by VLC using petroleum ether-CHCl₃ mixtures (20 and 50%) as solvent system. Subfraction 7 after treatment with CH₂N₂ afforded a complex mixture (535 mg) that was purified further by flash CC eluted with CHCl₃-MeOH mixtures (2, 5, 10, 20 and 50%), and prep. TLC [hexanes-MeOH (8:2)] gave methyl coronalolate acetate (**4**, 1.2 mg, 0.00008%).

The dried, powdered leaves of *G. sootepensis* (8.67 kg) were extracted with hexanes (3 x 20 L), CHCl₃ (3 x 20 L), EtOAc (3 x 20 L), and MeOH (4 x 20 L) at room temperature. Removal of solvents from each extract under vacuum afforded greenish residues (369.7, 535.8, 212.6 and 191.3 g, respectively). The active chloroform extract (348 g, P-388 murine lymphocytic leukemia cell line, ED₅₀ 1.6 µg/ml) was subjected to flash CC (Si gel H) using hexanes and hexanes-EtOAc mixtures of increasing polarity, and MeOH for elution. Fraction 8 (74 g) was purified by flash CC using hexanes, CH₂Cl₂ and MeOH mixtures of increasing polarity to afford a gum (17.6 g) which was crystallized to yield **1** (3.8 g) as colorless plates upon the addition of hexanes-CH₂Cl₂ mixture. Fraction 9 (129 g) was repeatedly chromatographed on Si gel 60 using hexanes, CH₂Cl₂ and MeOH mixtures of increasing polarity to afford **2** (2.3 g).

Coronalolide methyl ester [3,4-seco-cycloart-4(28),24-diene-6-hydroxy-26-ol-3,29-dioic acid-29- α , γ -lactone-3-methyl ester] (**1**). Colorless needles from MeOH: mp 91-92.5°; $[\alpha]_D^{25} +121.6^\circ$ (*c* 0.86, CHCl₃); UV $\lambda_{\max}^{\text{MeOH}}$ nm: 213 and 229; IR ν_{\max} (film) cm⁻¹: 2943, 1768 (γ -lactone), 1739 (COOR), 1684 (CHO), 1641 (C=CH₂), 1265 and 1167; ¹H-NMR: Table 1; ¹³C-NMR: Table 2; EIMS *m/z* (rel. int. %): 496 (14) [M]⁺, 481 (9) [M-CH₃]⁺, 371 (46) [M-C₈H₁₃O]⁺, 231 (18), 217 (23), 157 (45), 145 (49), 129 (68), 125 (24), 105 (100); HR-EIMS: *calcd* for C₃₁H₄₄O₅ 496.3189, found 496.3182.

Coronalolide (**2**). White powder from hexanes-CH₂Cl₂: mp 82.5-83°; $[\alpha]_D^{25} +119.1^\circ$ (*c* 0.69, CHCl₃); UV $\lambda_{\max}^{\text{MeOH}}$ nm: 241 and 268; IR ν_{\max} (film) cm⁻¹: 3221 (broad shoulder, COOH), 2946, 2880, 1761 (γ -lactone), 1707 (COOH), 1686 (CHO), 1641 (C=CH₂), 1458, 1268, 1148, 1028, 988 and 757; ¹H-NMR: Table 1; ¹³C-NMR: Table 2; EIMS *m/z* (rel. int. %): 482 (25) M⁺, 464 (15) [M-H₂O]⁺, 384 (22), 357 (100) [M-C₈H₁₃O]⁺, 245 (13), 219 (18), 147 (37), 121 (41), 107 (49); HR-EIMS: *calcd* for C₃₀H₄₂O₅ 482.3032, found 482.3027.

Coronalolic acid [3,4-seco-cycloart-4(28),24-dien-26-ol-29-hydroxy-3-oic acid] (**3**). Amorphous gum: $[\alpha]_D^{20} -36.4^\circ$ (*c* 0.21, CHCl₃); UV $\lambda_{\max}^{\text{MeOH}}$ nm: 232; IR ν_{\max} (film) cm⁻¹: 3431 (OH), 2930, 1707 (COOH), 1688 (CHO), 1456, 1377, 1177, 1059, 988 and 901; ¹H-NMR: Table 1; ¹³C-NMR: Table 2; EIMS *m/z* (rel. int. %): 470 (36) M⁺, 455 (100), 452 (39) [M-H₂O]⁺, 437 (55), 424 (38), 345 (20) [M-C₈H₁₃O]⁺, 327 (30), 285 (18), 233 (54), 147 (84), 109 (86), 105 (90); HR-EIMS: *calcd* for C₃₀H₄₆O₄ 470.3396, found 470.3405.

Methyl coronalolate acetate (4). Amorphous gum: $^1\text{H-NMR}$ (500 MHz, CDCl_3): Table 1; CIMS m/z (rel. int. %): 527 (42) $[\text{M}+1]^+$, 485 (14) $[\text{M}+1-\text{CH}_2=\text{C}=\text{O}]^+$, 467 (100) $[\text{M}+1-\text{AcOH}]^+$.

X-ray experimental data and structure analysis of 1: *Crystal data.* $\text{C}_{31}\text{H}_{44}\text{O}_5$, $M_r=496.66$, Monoclinic, $P2_1$, $a=7.413(4)$, $b=27.85(2)$, $c=7.486(5)$ Å, $\beta=116.44(5)^\circ$, $V=1384(2)$ Å³ (by least squares refinement on diffractometer angles for 12 automatically centered reflections), $\lambda=1.54178$ Å, $Z=2$, $D_c=1.192$ Mg/m³, $F(000)=540$, $\mu(\text{Cu-K}\alpha)=0.627$ mm⁻¹. *Data Collection and Processing.* Three-dimensional, room temperature (293°K) X-ray data were collected on a Rigaku AFC6S diffractometer with monochromatized Cu-K α X-radiation. $2\theta/\omega$ mode with scan range (ω) 6.36-57.46° plus K α separation and a variable scan speed (4.88-14.65 min⁻¹). A total of 2110 reflections was measured ($3 < 2\theta < 115^\circ$, min. hkl -8 0 0, max. hkl 7 30 8); 1945 independent reflections were obtained [$R(\sigma)=0.0319$, Friedel opposites merged]. No absorption correction was applied. One control reflection monitored every 99 reflections showed no appreciable decay during 13.6 h of exposure of the crystal to X-rays. *Structure Analysis and Refinement.* Direct methods resulted in the location of all of the non-hydrogen atoms. Full matrix least-squares refinement with anisotropic thermal parameters was used for all non-hydrogen atoms. Hydrogen atoms were refined in riding mode. Refinement converged at $R=0.0463$, $R_w=0.1228$. Maximum and mean shift/error in the final cycle of refinement were 0.006 and 0.003, respectively. The final electron-density difference synthesis showed no peaks >0.16 or <-0.14 eÅ⁻³. All computations were carried out using the SHELXTL for IRIS V5.03 system of programs.²⁸⁾

Bioassay evaluations: Compounds 1-3 were screened for cytotoxicity against a panel of human cancer cell lines, according to established protocols.²⁵⁾ ED_{50} values of >4 µg/ml were regarded as inactive.

Acknowledgments: This investigation was supported by grant U01-CA52956 from the National Cancer Institute, NIH, Bethesda, Maryland. G.L.S. thanks CONICET-Argentina for a fellowship. We wish to acknowledge Mr. R.B. Dvorak, University of Illinois at Chicago, for the mass spectra and the Research Resources Center, University of Illinois at Chicago for the provision of NMR facilities. The Thai team wishes to acknowledge the financial support from the National Research Council of Thailand. The provision of high-resolution NMR spectra by Prof. Walter C. Taylor and Dr. Lindsay Byrne, of the University of Sydney and the University of Western Australia, Australia, respectively, are gratefully acknowledged.

REFERENCES AND NOTES

1. Woo, W.S.; Lee, E.B.; Shin, K.H.; Kang, S.S.; Chi, H.J. *Kor. J. Pharmacog.* **1981**, *12*, 153-170.

2. Mbela, T.K.M.; Shabani, M.; Dieyi, S.; Cimanga, K.; Moswa, L. *Fitoterapia* **1992**, *63*, 179-181.
3. Jain, S.P. *Int. J. Crude Drug Res.* **1989**, *27*, 29-32.
4. El-Hamidi, A. *Planta Med.* **1970**, *18*, 278-280.
5. Croft, K.D.; Tu'Ipulotu, P. *S. Pacific J. Nat. Sci.* **1980**, *1*, 45-47.
6. Nayar, S.L. *Bull. Natl. Inst. Sci. India* **1955**, *4*, 137-145.
7. Lu, R.F.; Aeng, Y.R.; Wang, W.C.; *Sheng Chih Yu Bi Yun* **1981**, *12*, 16-18.
8. Laurens, A.; Mboup, S.; Tignokpa, M.; Sylla, O.; Masquelier, J. *Pharmazie* **1985**, *40*, 482-485.
9. Takase, H.; Imanishi, K.; Miura, O.; Yumioka, E.; Watanabe, H. *Jap. J. Pharmacol.* **1989**, *49*, 301-308.
10. Hussain, M.M.; Sokomba, E.N.; Shok, M. *Int. J. Pharmacog.* **1991**, *29*, 94-100.
11. Manson, D. *J. Malaria Inst. India* **1939**, *2*, 85-93.
12. Kinghorn, A.D.; Farnsworth, N.R.; Beecher, C.W.W.; Soejarto, D.D.; Cordell, G.A.; Pezzuto, J.M.; Wall, M.E.; Wani, M.C.; Brown, D.M.; O'Neill, M.J.; Lewis, J.A.; Besterman, J.M. *Int. J. Pharmacog.* **1995**, *33* (Suppl.), 48-58.
13. Kikuchi, M.; Yoshikoshi, A. *Chem. Lett.* **1972**, 725-728.
14. Sun, H.-D.; Qiu, S.-X.; Lin, L.-Z.; Wang, Z.-Y.; Lin, Z.-W.; Pengsuparp, T.; Pezzuto, J.M.; Fong, H.H.S.; Cordell G.A.; Farnsworth, N.R. *J. Nat. Prod.* **1996**, *59*, 525-527.
15. Liu, J.-S.; Huang, M.-F.; Ayer, W.A.; Bigam, G. *Tetrahedron Lett.* **1983**, *24*, 2355-2358.
16. Raldugin, V.A.; Kukina, T.P.; Yaroshenko, N.I.; Pentegora, V.A. *Khim. Prir. Soedin.* **1987**, *2*, 306-307.
17. Chen, Y.; Lin, Z.; Zhang, H.; Sun, H. *Phytochemistry* **1990**, *29*, 3358-3359.
18. Tan, R.; Xue, H.; Li, L.-N. *Planta Med.* **1991**, *57*, 87-88.
19. Chen, G.F.; Li, Z.L.; Pan, D.J.; Tang, C.M.; He, X.; Xu, G.Y.; Chen, K.; Lee, K.H. *J. Nat. Prod.* **1993**, *56*, 1114-1122.
20. Cabrera, G.M.; Gallo, M.; Seldes, A.M. *J. Nat. Prod.* **1996**, *59*, 343-347.
21. Herz, W. *Isr. J. Chem.* **1977**, *16*, 32-44, and references therein.
22. Achenbach, H.; Frey, D. *Phytochemistry* **1992**, *31*, 4263-4274.
23. Haasnoot, C.A.G.; De Leeuw, F.A.A.M., Altona, C. *Tetrahedron* **1980**, *36*, 2783-2792.
24. Cabrera, G.M.; Gallo, M.; Seldes, A.M. *Phytochemistry* **1995**, *39*, 665-666.
25. Likhitwitayawuid, K.; Angerhofer, C.K.; Cordell, G.A.; Pezzuto, J.M.; Ruangrunsi, N. *J. Nat. Prod.* **1993**, *56*, 30-38.
26. Liu, Y.-L.; Mabry, T.J. *Phytochemistry* **1981**, *20*, 309-311.
27. Cook, I.B. *Aust. J. Chem.* **1989**, *42*, 1493-1518.
28. Sheldrick, G.M. SHELXTL for IRIS V5.03 (Copyright 1990-1995, SAXI, Siemens Analytical X-ray Instruments, Inc.).